

Total Synthesis of Kopsinine

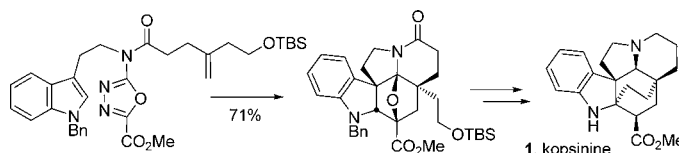
Jian Xie, Amanda L. Wolfe, and Dale L. Boger*

Department of Chemistry and The Skaggs Institute for Chemical Biology,
The Scripps Research Institute, 10550 North Torrey Pines Road, La Jolla,
California 92037, United States

boger@scripps.edu

Received December 31, 2012

ABSTRACT



The use of a powerful intramolecular [4 + 2]/[3 + 2] cycloaddition cascade of an 1,3,4-oxadiazole in the divergent total synthesis of kopsinine (**1**), featuring an additional unique SmI_2 -promoted transannular cyclization reaction for formation of the bicyclo[2.2.2]octane central to its hexacyclic ring system, is detailed.

In recent efforts that targeted key members of the *Aspidosperma* alkaloids, including minovine,¹ (+)-fendleridine (aspidospermidine),² (–)-aspidospermine and (+)-spiegazzinine,³ (+)-*N*-methylassidospermidine, (–)-vindorosine and (–)-vindoline,⁴ and their extension to the total synthesis of vinblastine⁵ and related natural products including vincristine,⁶ and key analogues,⁷ we developed a powerful intramolecular [4 + 2]/[3 + 2] cycloaddition cascade of 1,3,4-oxadiazoles that provides the pentacyclic

core and all the stereochemistry of the natural products in a single step.⁸ Herein, we report the extension of these studies to the total synthesis of kopsinine (**1**),⁹ a *Kopsia* alkaloid first isolated from *Kopsia Longiflora* Merr. and related to the *Aspidosperma* alkaloids by virtue of an additional bond formed by joining the terminal methyl group (C21) of the C5 ethyl substituent with C2 to provide a bicyclo[2.2.2]octane central to its hexacyclic ring system (Figure 1). To date, this bicyclo[2.2.2]octane central to the *Kopsia* core has been accessed only by key Diels–Alder reactions of unnatural aspidosperma-like pentacyclic dienes (C2–C5) necessarily lacking a C5 substituent.¹⁰ Complementary to these efforts, the total synthesis detailed herein enlists a late stage C2–C21 bond formation in an approach that directly links **1** to the structures of the corresponding *Aspidosperma* alkaloids. The added bonus of the strategy is that

(1) Yuan, Z. Q.; Ishikawa, H.; Boger, D. L. *Org. Lett.* **2005**, *7*, 741.
(2) Campbell, E. L.; Zuhl, A. M.; Liu, C. M.; Boger, D. L. *J. Am. Chem. Soc.* **2010**, *132*, 3009.

(3) Lajiness, J. P.; Jiang, W.; Boger, D. L. *Org. Lett.* **2012**, *14*, 2078.
(4) (a) Wolkenberg, S. E.; Boger, D. L. *J. Org. Chem.* **2002**, *67*, 7361.
(b) Choi, Y.; Ishikawa, H.; Velcicky, J.; Elliott, G. I.; Miller, M. M.; Boger, D. L. *Org. Lett.* **2005**, *7*, 4539. (c) Elliott, G. I.; Velcicky, J.; Ishikawa, H.; Li, Y.; Boger, D. L. *Angew. Chem., Int. Ed.* **2006**, *45*, 620.
(d) Ishikawa, H.; Elliott, G. I.; Velcicky, J.; Choi, Y.; Boger, D. L. *J. Am. Chem. Soc.* **2006**, *128*, 10596. (e) Ishikawa, H.; Boger, D. L. *Heterocycles* **2007**, *72*, 95. (f) Kato, D.; Sasaki, Y.; Boger, D. L. *J. Am. Chem. Soc.* **2010**, *132*, 3685. (g) Sasaki, Y.; Kato, D.; Boger, D. L. *J. Am. Chem. Soc.* **2010**, *132*, 13533.

(5) (a) Ishikawa, H.; Colby, D. A.; Boger, D. L. *J. Am. Chem. Soc.* **2008**, *130*, 420. (b) Gotoh, H.; Sears, J. E.; Eschenmoser, A.; Boger, D. L. *J. Am. Chem. Soc.* **2012**, *134*, 13240.

(6) Ishikawa, H.; Colby, D. A.; Seto, S.; Va, P.; Tam, A.; Kakei, H.; Rayl, T. J.; Hwang, I.; Boger, D. L. *J. Am. Chem. Soc.* **2009**, *131*, 4904.

(7) (a) Va, P.; Campbell, E. L.; Robertson, W. M.; Boger, D. L. *J. Am. Chem. Soc.* **2010**, *132*, 8489. (b) Tam, A.; Gotoh, H.; Robertson, W. M.; Boger, D. L. *Bioorg. Med. Chem. Lett.* **2010**, *20*, 6408. (c) Gotoh, H.; Duncan, K. K.; Robertson, W. M.; Boger, D. L. *ACS Med. Chem. Lett.* **2011**, *2*, 948. (d) Leggans, E. K.; Barker, T. J.; Duncan, K. K.; Boger, D. L. *Org. Lett.* **2012**, *14*, 1428. (e) Leggans, E. K.; Duncan, K. K.; Barker, T. J.; Schleicher, K. D.; Boger, D. L. *J. Med. Chem.* **2013**, DOI: 10.1021/jm3015684. (f) Schleicher, K. D.; Sasaki, Y.; Tam, A.; Kato, D.; Duncan, K. K.; Boger, D. L. *J. Med. Chem.* **2013**, *56*, 483.

(8) (a) Wilkie, G. D.; Elliott, G. I.; Blagg, B. S. J.; Wolkenberg, S. E.; Soenen, D. R.; Miller, M. M.; Pollack, S.; Boger, D. L. *J. Am. Chem. Soc.* **2002**, *124*, 11292. (b) Elliott, G. I.; Fuchs, J. R.; Blagg, B. S. J.; Ishikawa, H.; Tao, H.; Yuan, Z. Q.; Boger, D. L. *J. Am. Chem. Soc.* **2006**, *128*, 10589.

(9) (a) Crow, W. D.; Michael, M. *Aust. J. Chem.* **1955**, *8*, 129. (b) Crow, W. D.; Michael, M. *Aust. J. Chem.* **1962**, *15*, 130. (c) Kump, W. G.; Schmid, H. *Helv. Chim. Acta* **1961**, *44*, 1503. (d) Kump, W. G.; Count, D. J. L.; Battersby, A. R.; Schmid, H. *Helv. Chim. Acta* **1962**, *45*, 854. (e) Kump, W. G.; Patel, M. B.; Rowson, J. M.; Schmid, H. *Helv. Chim. Acta* **1964**, *47*, 1497.

(10) (a) Kuehne, M. E.; Seaton, P. J. *J. Org. Chem.* **1985**, *50*, 4790. (b) Ogawa, M.; Kitagawa, Y.; Natsume, M. *Tetrahedron Lett.* **1987**, *28*, 3985. (c) Wenkert, E.; Pestchanker, M. J. *J. Org. Chem.* **1988**, *53*, 4875. (d) Magnus, P.; Brown, P. *J. Chem. Soc., Chem. Commun.* **1985**, 184. (e) Jones, S. B.; Simmons, B.; Mastracchio, A.; MacMillan, D. W. C. *Nature* **2011**, *475*, 183. (f) Harada, S.; Sakai, T.; Takasu, K.; Yamada, K.; Yamamoto, Y.; Tomioka, K. *Chem. Asian J.* **2012**, *7*, 2196.

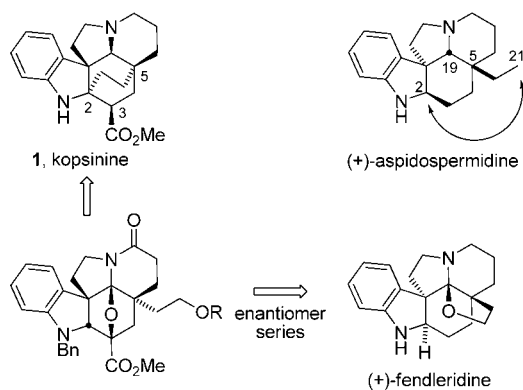


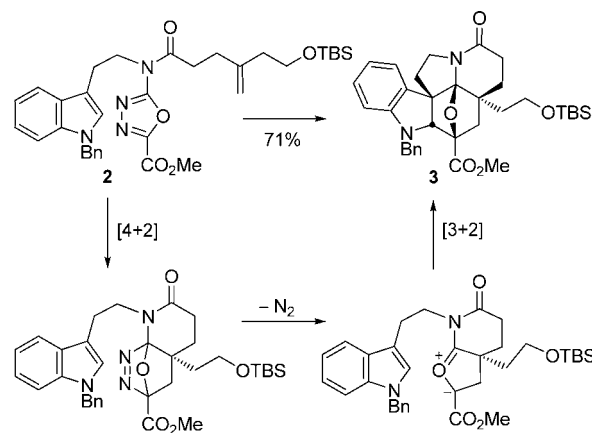
Figure 1. Structure of kopsinine, relationship to *Aspidosperma* alkaloids, and late stage divergent total syntheses of kopsinine (**1**) in an approach fundamentally distinct from prior reports.

late stage modification of a route developed to access (+)-fendleridine² now permits a divergent¹¹ synthesis of kopsinine (**1**) in an approach fundamentally distinct from prior reports.

As detailed in earlier studies, the key intramolecular [4 + 2]/[3 + 2] cycloaddition was accomplished upon warming a solution of **2** at 180 °C in *o*-dichlorobenzene (*o*-DCB) to provide **3** as a single diastereomer in yields as high as 71% (Scheme 1).² The cycloaddition cascade is initiated by an intramolecular Diels–Alder reaction of the tethered unactivated dienophile with the central 1,3,4-oxadiazole.^{12,13} Loss of N₂ from the initial cycloadduct generates a 1,3-dipole that is uniquely stabilized by the complementary substitution at each of the dipole termini. The ensuing 1,3-dipolar cycloaddition reaction proceeds with a regioselectivity that is dictated by the linking tether, but is reinforced by the intrinsic polarity of the reacting partners, and with a diastereoselectivity that is derived from an indole endo [3 + 2] cycloaddition, in which the dipolarophile is sterically directed to the face opposite the newly formed six-membered ring.^{8,14} Four C–C bonds, three rings, five stereocenters, and a pentacyclic skeleton are assembled in a single transformation.

Reductive oxido bridge opening of **3** was accomplished upon treatment with NaCNBH₃ in 20% HOAc/*i*-PrOH to provide the alcohol **4** (87%) as a single diastereomer, resulting from convex face hydride reduction of an intermediate *N*-acyliminium ion that is flanked by two quaternary centers (Scheme 2). Conversion of **4** to the methylthiocarbonate **5** (NaH, CS₂, THF, 0 °C, 1 h followed by MeI, 25 °C, 1 h, 78%) set the stage for a Chugaev elimination.¹⁵ Warming a solution of the xanthate **5** in *o*-dichlorobenzene (*o*-DCB) at 150 °C for 1 h provided a

Scheme 1



separable 2:1 mixture of **7** and **6** in excellent yield 85%. The observation of a small amount of the rearranged and stable *S*- versus *O*-dithiocarbonate could be avoided by conducting the reaction at a lower temperature in benzene in a sealed vessel at a bath temperature of 130 °C (6 h), affording the elimination products in superb yield (95%). Prior to continuing to address improvements in the regioselectivity of the elimination reaction and with sufficient **6** in hand, studies on the key formation of the C2–C21 bond were conducted. Deprotection of the TBS ether (Bu₄NF, THF, 98%) in **6** and conversion of the primary alcohol **8** to the methylthiocarbonate **9** (86%) set the stage for the key bond formation. Treatment of **9** (0.1 M) with SmI₂ in 10:1 THF–HMPA (25 °C, 20 min) provided **10** in excellent yield (75%) as a single diastereomer, presumably resulting from a radical-mediated cyclization followed by kinetic protonation of the further reduced conjugate addition ester enolate from the less hindered convex face.¹⁶

Although not investigated herein, Molander and co-workers have shown that the scope of such SmI₂-mediated conjugate addition reactions is most consistent with radical-mediated versus anionic cyclization.¹⁶ Consistent with this expectation, a (TMS)₃SiH-mediated free radical cyclization (C₆H₆, 90 °C, 2 h) of **9** also provided the analogous addition product in good yield (60%), but as a 1:1 mixture of C3 diastereomers. Also consistent with the observations of Molander, the product **10**, obtained as the exclusive diastereomer from the SmI₂-mediated reaction, represents the less stable of the two C3 diastereomers ($\Delta E = 1.7$ kcal/mol), indicating that it is the result of kinetic versus thermodynamic protonation of the enolate derived from a subsequent SmI₂ reduction of the radical addition intermediate. Completion of the concise total synthesis simply involved treatment of **10** with Lawesson's reagent¹⁷ to provide the thiolactam **11** (90%) which upon treatment with Raney-Ni (EtOH, 25 °C, 3 h, 95%) underwent both

(11) Boger, D. L.; Brotherton, C. E. *J. Org. Chem.* **1984**, *49*, 4050.

(12) (a) Boger, D. L. *Tetrahedron* **1983**, *39*, 2869. (b) Boger, D. L. *Chem. Rev.* **1986**, *86*, 781.

(13) Margetic, D.; Troselj, P.; Johnston, M. R. *Mini-Rev. Org. Chem.* **2011**, *8*, 49.

(14) (a) Padwa, A.; Price, A. T. *J. Org. Chem.* **1995**, *60*, 6258.

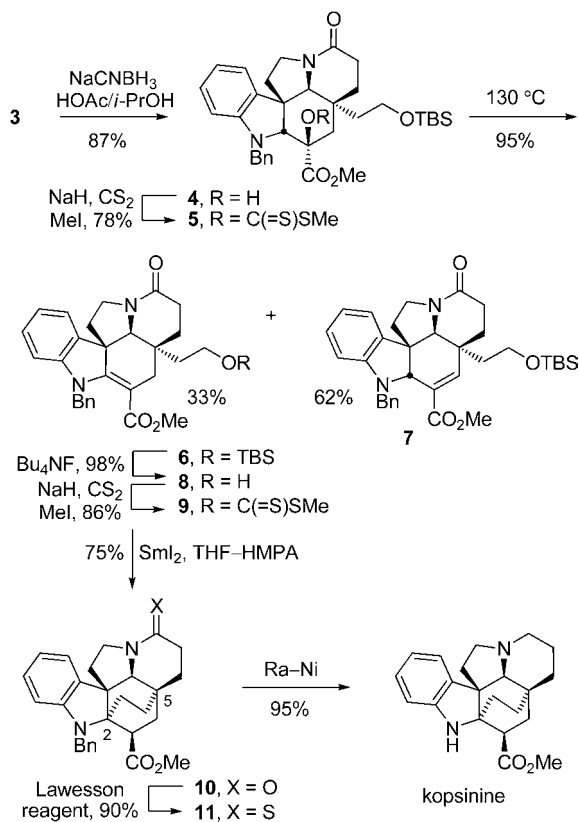
(b) Padwa, A.; Price, A. T. *J. Org. Chem.* **1998**, *63*, 556.

(15) DePuy, C. H.; King, R. W. *Chem. Rev.* **1960**, *60*, 431.

(16) Molander, G. A.; Harris, C. R. *J. Org. Chem.* **1997**, *62*, 7418.

(17) Yde, B.; Yousif, N. M.; Pedersen, U.; Thomsen, I.; Lawesson, S. O. *Tetrahedron* **1984**, *40*, 2047.

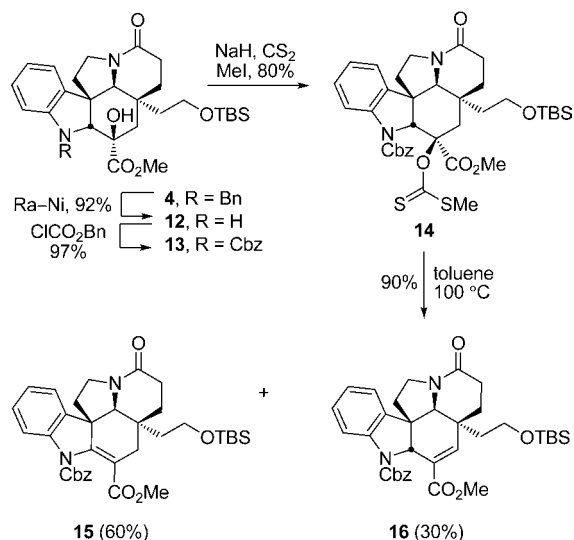
Scheme 2



desulfurization and *N*-debenzylation to provide kopsinine (**1**), spectroscopically (^1H and ^{13}C NMR) identical to authentic material.^{10e}

In efforts that improved the regioselectivity of the Chugaev elimination, conversion of **5** to the corresponding indoline Cbz carbamate **13**, followed by methyl dithiocarbonate formation provided **14** (Scheme 3). The intermediate xanthate **14** underwent the thermal elimination reaction under much milder reaction conditions (toluene, 100 °C, 48 h) than **5**, providing a superb yield (90%) of the separable elimination products **15** and **16** in a reversed 2–2.7:1 ratio favoring the $\Delta^{2,3}$ isomer **15** (60%). Clearly the amine carbamate substitution activates C2–H for xanthate syn elimination, now favoring formation of the

Scheme 3



more substituted and stable olefin. Although not examined, the effective access to **15** provides, in principle, the development of an improved approach to kopsinine.

Herein, we reported the use of a powerful intramolecular [4 + 2]/[3 + 2] cycloaddition cascade of 1,3,4-oxadiazoles in the divergent total synthesis of kopsinine (**1**), featuring an additional unique SmI_2 -promoted transannular cyclization reaction for formation of the bicyclo[2.2.2]octane central to its hexacyclic ring system and directly linking it with the pentacyclic *Aspidosperma* alkaloids to which it is related.

Acknowledgment. We gratefully acknowledge the financial support of the National Institutes of Health (CA042056) and the Skaggs Institute for Chemical Biology. A.L.W. is a Skaggs Fellow and recipient of a NSF graduate fellowship (2009–2012).

Supporting Information Available. Full experimental details, compound characterizations, and spectra are provided. This material is available free of charge via the Internet at <http://pubs.acs.org>.

The authors declare no competing financial interest.